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TECHNICAL NOTE

# Experimental Study on TGA, XRD and SEM Analysis of Concrete with Ultra-fine Slag

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PAPER INFO

ABSTRACT

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Keywords: Alccofine Thermogravimetric Analysis X-ray diffraction Scanning Electron Microscope The performances of cementitious materials as well as the efficiency of construction are adversely affected at high temperatures. Previous studies have already demonstrated that ultra-fine (alccofine) material accelerates the hydration of cement particles and subsequently improves the mechanical and durability properties of the concrete at normal temperature. Moreover, at higher temperatures the performance of the concrete with the addition of alccofine is still unknown. This paper presents the effect of analytical properties of concrete with alccofine (25%) as a replacement of cement for various W/B ratios (i.e. 0.38, 0.4 and 0.45). The effect of addition of alccofine dosage on the thermogravimetric analysis was carried out in which not only mass loss and decomposition of hydration products from concrete with respect to temperature was found but also bound water (%) and calcium hydroxide (%) present in the concrete were determined. From X-ray diffraction it was observed that alccofine improved the formation of calcium silicate hydrate and calcium silicate aluminum hydrate in concrete and scanning electron microscope (SEM) analysis showed the formation of ettringite needles and calcium silicate hydrate in voids which made concrete denser. Therefore, it was concluded from this study that alccofine can be used as a viable substitute to cement in normal concrete considering its positive effects on property enhancement and an eco-friendly product.

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### 1. INTRODUCTION

The most regularly utilized building material is concrete, generally made from Portland cement [1]. Every material utilized for making concrete affects the environment and gives rise to sustainability issues [2]. However, the manufacture of Portland cement creates a problem such as emission of CO<sub>2</sub> (approximately 7 to 8% of total man-made) to the atmosphere and global warming [3]. By the addition of a few alternative (pozzolanic) materials, the diverse properties of concrete i.e. strength, workability, durability, and permeability may be improved [4]. Researchers have developed different cementing material which can be used fully or partially to replace cement which will decrease the emission of greenhouse gases [5]. These alternative materials can be added to concrete mixes as

replacement of cement or aggregate depending on their physical and chemical properties [6].

According to available literatures, the replacement of cement by industrial by-products which possess pozzolanic nature such as red mud, fly ash [7,8], metakaolin, silica-fume [9], rice-husk ash [10], mine waste, granulated blast furnace slag, etc., have shown improved results than the conventional concrete in terms of fresh, mechanical properties and durability [11]. Recently, some of the researchers have reported that the ultra-fine material namely alcoofine (ALC) [12-14], obtained as a by-product from iron ore industry in India also possess pozzolanic nature and can be used as a fully or partial replacement by weight of cement in concrete.

Chopra and Siddique [15] have studied the effect of rush husk ash (RHA) on strength, durable and microstructural properties were studied. At 15% RHA replacement showed huge formation of calcium silicate hydrate and steady state hydration were obtained. Other investigators have studied the microstructural,

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hydration, mechanical properties with metakaolin. The portlandite changes into calcium silicate hydrate showed better formation by XRD analysis at 15% of metakaolin in the concrete. Rathore and Chawda [16] have done an experimental investigation on high-performance concrete with the replacement of sand by M-sand and partial replacement of cement by alcoofine and fly ash for M60 grade of concrete. From the study, they conformed that the strength improvement in concrete with alcoofine is higher than that of fly ash. Gayathri et al. [17] conducted research on the performance of concrete with partial replacement with alcoofine by weight of cement for M30 grade of concrete and she observed improvement in the strength of concrete at 15% replacement.

The effect of alcoofine on the durability and strength properties of the conventional concrete has been investigated [13] but its effects on the thermogravimetric analysis, X-ray diffraction analysis and scanning electronic microscopy analysis of the concrete with alcoofine have not been well investigated. Therefore, this study was designed to assess the effectiveness of alcoofine as constant replacement to the cement on the aforementioned properties of mass concrete.

#### 2. RESEARCH SIGNIFICANCE

The Ordinary Portland Cement, which is a major constituent of conventional concrete, plays a significantly important role in attaining the strength properties. However, now-a-days manufacturing of cement had become a major source for pollution, which propels the researchers to replace cement by some alternative pozzolanic materials that can provide the desirable mechanical and durability properties to concrete as well as address the pollution menance. Previous researchers focused on the use of alccofine in concrete mix to improve the mechanical and durability properties of concrete. In the present study, an attempt has been made to study the effect of alcoofine as partial replacement of cement on analytic properties of concrete so that their scope to address on thermal and microstructural properties can be explored.

#### 3. MATERIALS

- **3. 1. Cement** In this investigation, OPC 43 grade was used and it was tested as per Indian standard specifications [12]. Figure 1 shows the SEM image of cement.
- **3. 2. Alccofine** AL-1203 was obtained from Ambuja Cement Ltd, Goa having the specific gravity of 2.9 confirming to ASTM C989-1999 was used in entire

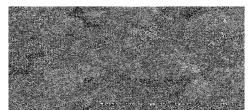


Figure 1. SEM image of cement

investigation. The physical and chemical properties of AL are given in Tables 1 and 2. Figures 2 and 3 are shown SEM and EDX images of alcoofine.

**TABLE 1.** Physical properties of Alccofine

Characteristics	Test Results					
Specific gravity	2.9					
Specific surface area [m²/kg]	1200					
Bulk density [kg/m³]	680					
Particle Size in Micron						
$D_{10}$	1.5					
$D_{50}$	5					
$D_{90}$	9					

**TABLE 2.** Chemical Properties of AL form EDAX

Results for EDAX			
Weight %	Atomic %		
45.69	57.64		
35.26	33.39		
4.01	2.25		
6.38	3.44		
8.66	3.27		
	Weight %  45.69  35.26  4.01  6.38		

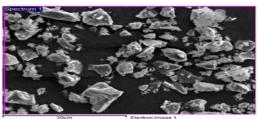


Figure 2. SEM image of Alccofine

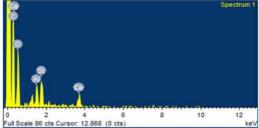


Figure 3. EDX image of alccofine

- **3. 3. Aggregates** Natural river sand with specific gravity 2.68 was used conforming to zone II as per Indian standard specification. The crushed stone angular aggregate was used as coarse aggregate with the specific gravity of 2.79. The maximum size of the aggregate was limited to 20 mm.
- **3. 4. Water** In this investigation, ordinary tap water available in campus was used for all concrete mixes.

#### 4. DATA, AND EXPERIMENTAL INVESTIGATION

- **4. 1. Mix Proportion** The ingredients of concrete i.e. Cement, coarse aggregate, and fine aggregate were mixed in the proper proportion by addition of water as shown in Table 3. Alcoofine was added to the mixes as a replacement of cement by 25% mass.
- **4. 2. Preparation of Sample** Mixing procedure and curing method was kept the same for TGA, XRD and SEM samples. For TGA and XRD, the samples were ground and sieved through 63 lm sieve. To stop hydration and to remove physically bound water the solvent exchange method was adopted using acetone. A 100 mL of acetone was added to 30 g of the sieved sample in a plastic bottle and mixed vigorously by hand for about 3 min. Excess acetone was drained out and the process was repeated. The samples were then dried overnight in an oven at 40 □ C temperature. The dried samples were collected and stored in a sealed plastic container till the time of testing.

#### 4.3. Methods

**4. 3. 1. Thermogravimetric analysis** Total of twelve samples (with and without alcofine) were studied by Thermogravimetric analysis. Small samples were taken from the concrete cube and heat was applied from 25°C to 1000°C in a thermal analyzer with a rate of heating 10°C/min. From TGA experiment, weight losses with respect to increasing temperature and peak temperature for the decomposition of hydrates like Ca (OH)<sub>2</sub>, calcite, calcium silicate hydrate and C-A-S-H.

**TABLE 3.** Mix proportions for trail mixes

S.No	M	C	AL	FA	CA	W/B
1	NM1	438.13	0	644.59	1118.42	0.45
2	AF1	328.59	109.53	644.59	1118.42	0.45
3	NM2	492.90	0	610.55	1106.06	0.40
4	AF2	369.67	123.22	610.55	1106.06	0.40
5	NM3	518.84	0	595.98	1098.62	0.38
6	AF3	388.18	155.65	595.98	1098.62	0.38

Units: M-Mix, C-Cement, AL-Alccofine, FA- Sand and CA-Gravel W/B-Water to binder ratio in kg/m<sup>3</sup>

- 4. 3. 2. X-Ray Diffraction Analysis (XRD) XRD is an important method used for the quantitative and qualitative analysis of hydrated cement concrete samples. The X-Ray diffraction is based on Bragg's law. By using X-Ray diffraction test results, a graph between the angle at which the wave was diffracted and intensity of X-ray was determined. The XRD test was conducted on samples on the 28th day. The diffractometer with CuKu source having a wavelength of  $\chi = 1.54 \text{\AA}$  at the scanning speed of 2s/ step and the diffraction angle between 7° and 60° was employed for test conduction.
- **4. 3. 3. Scanning Electronic Microscope Analysis** (SEM analysis) SEM analysis was used to determine the microstructure of the hardened concrete with and without alcofine. The concrete samples were taken in a small size of 5 mm, which consist of fine aggregate and mortar.

#### 5. RESULTS AND DISCUSSION

thermogravimetric Analysis From thermogravimetric analysis results, the mass loss along with the decomposition of all components with temperature was found. The thermogravimetric analysis showed that significant weight loss can occur in many ways. The primary effect from 25°C to 100°C had to do with the residual pore water which evaporated from capillary pores [13]. In this stage, the weight loss depended upon the adsorbed water, interlayer water and capillary pores. The second effect from 100 to 450°C had to do with dehydration of calcium silicate hydrates, ettringite and C-A-H. The third at about 475-750°C is due to the decomposition of calcium hydroxide formed during hydration as shown in Equation 1 [18].

$$Ca(OH)_2 \rightarrow CaO + H_2O$$
 (1)

The last weight loss effect around 800°C can be attributed to decarbonization of CaCO<sub>3</sub> as shown in Equations (2) and (3).

$$Ca(OH)_2 + CO_2 \rightarrow CaCO_3 + H_2O$$
 (Carbonation) (2)

$$CaCO_3 \rightarrow CaO + CO_2$$
 (Decarbonation) (3)

Figure 4 shows the decomposition of all hydration compounds in the temperature range of 23-800°C and Figure 5 shows the weight loss of normal and high dosage alcoofine concrete at different water to binder ratios with respect to temperature [14]. It can be clearly seen from Figure 5 that the percentage of weight loss in alcoofine concrete is less than that of the normal concrete. From Equations 4 and 5, the content of calcium hydroxide and bound water were calculated as a percentage of the weight at 580°C [19] as follows:

Calcium hydroxide (%) = 
$$\frac{W_{400} - W_{580}}{W_{580}} \times \frac{74}{18} \times 100$$
 (4)

The molecular weight of Calcium hydroxide is 74 and that of water is 18.

Bound water (%) = 
$$\frac{W_{26} - W_{580}}{W_{580}} \times 100$$
 (5)

From obtained results percentage of calcium hydroxide was more in normal concrete but the percentage of bound water had more values in high dosage alccofine concrete compared to normal concrete [20] as shown in Figures 6 and 7. The percentage of calcium hydroxide was less in alccofine concrete because alccofine reacts with calcium hydroxide to form additional CSH gel whereas no such entity exists in normal concrete to react with the same thereby increasing the percentage of calcium hydroxide in normal concrete [21].

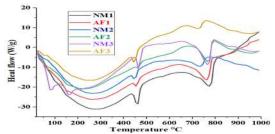


Figure 4. DSC curves of the concrete for different water to binder ratios

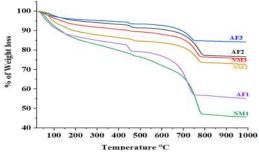
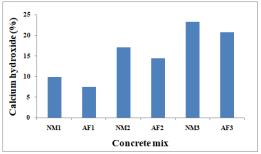


Figure 5. Percentage of weight loss curves of the concrete for different W/B ratio



**Figure 6.** Calcium hydroxide (%) of the concrete for various W/B ratio

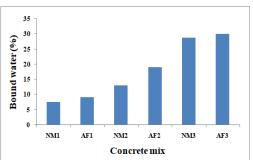


Figure 7. Bound water (%) of the concrete for various W/B

**5. 2. X-Ray Diffraction Analysis (XRD)** XRD analysis is a one of the method to find mineral composition like calcium silicate hydrate (C-S-H), calcium aluminosilicate hydrate (C-A-S-H), portlandite (Ca(OH)<sub>2</sub>) and ettringite present in the concrete [22]. Figures 8, 9 and 10 shows the peak intensities for control and alcoofine concrete at different water to binder ratio (i.e. 0.38, 0.4 and 0.45).

The intensity of calcium hydroxide  $(Ca(OH)_2)$  in the presence of alcofine at  $2\theta$ = 20.64 has increased compared to normal concrete for all water to binder ratios. Formation of  $Ca(OH)_2$  is more in normal concrete compared to alcofine concrete [14]. Further, the intensity of ettringite in the presence of alcofine at  $2\theta$ =22.87 has decreased compared to without alcofine. The peak intensity of calcium silicate hydrate at  $2\theta$ =26.46 has increased more in the presence of alcofine compared to normal concrete and the intensity of calcium aluminosilicate hydrate is more in normal concrete at  $2\theta$ =28.29 indicating rapid hydration. The formation of CSH is more in with high dosage alcofine concrete due to  $Ca(OH)_2$  converted to secondary CSH and therefore enhancing strength of the concrete [18-24].

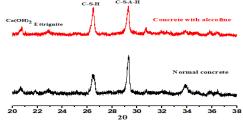
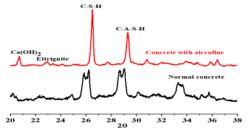
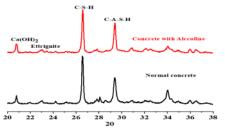


Figure 8. XRD analysis of the concrete at W/B = 0.45



**Figure 9.** XRD analysis of the concrete at W/B = 0.40

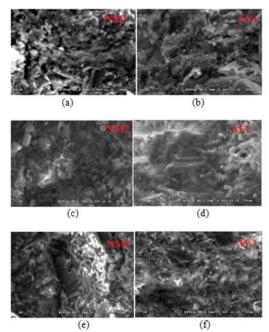


**Figure 10.** XRD analysis of the concrete at W/B = 0.38

**5. 3. Scanning Electronic Microscope Analysis** SEM images give information regarding the microstructure and surface morphology of calcium silicate hydrate and ettringite in concrete. It was observed that alcofine concrete gives less porous structure due to the formation of needles in voids compared to normal concrete at all water to binder ratios. With the addition of alcofine, concrete gives maximum strength compared to normal concrete due to full compact packing as a result of the filling of voids by alcofine [14]. Figure 11 SEM images of normal and high dosage alcofine concrete for different water to binder ratios.

#### 6. CONCLUSION

In this research, the effects of alcoofine on the thermal and microstructural properties of concrete were studied.



**Figure 11.** (a) Normal concrete at W/B is 0.45 (b) alcofine concrete at W/B is 0.45 (c) Normal concrete at W/B is 0.4 (d) alcofine concrete at W/B is 0.4 (e) Normal concrete at W/B is 0.38 (f) alcofine concrete at W/B is 0.38

From the thermogravimetric analysis, alcofine concrete was found to have more bound water and less calcium hydroxide (due to pozzolanic action) with respect to normal concrete at different water to binder ratio. It was also found that alcofine concrete had less weight loss and decomposition of hydrates compared to normal concrete thereby making it more stable. From XRD and SEM analysis, alcofine decreased the porous microstructure and improved the formation of calcium silicate hydrate and calcium silicate aluminum hydrate in the concrete. Hence, from the results obtained in this study it can be concluded that substituting the cement in concrete by alcofine is a feasible mean considering enhanced properties and eco-friendly nature of concrete.

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# Experimental Study on TGA, XRD and SEM Analysis of Concrete with Ultra-Fine Slag TECHNICAL NOTE

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Keywords: Alccofine Thermogravimetric Analysis X-ray diffraction Scanning Electron Microscope عملکرد مواد سیمان و همچنین بهره وری ساخت و ساز در دمای بالا تحت تأثیر قرار می گیرند. مطالعات قبلی قبلا نشان داده است که مواد فوق سنگین (آلکوفین) باعث سرعت بخشیدن به هیدراتاسیون ذرات سیمان و سپس بهبود خواص مکانیکی و دوام بتن در دمای طبیعی می شوند. علاوه بر این، در دماهای بالاتر عملکرد بتن با اضافه کردن آلکوفین هنوز ناشناخته است. این مقاله تاثیر خواص تحلیلی بتن را با آلکوفین (۲۵٪) به عنوان جایگزینی سیمان برای نسبت های مختلف ) B / Wیعنی ۲۰٫۸، ۶٫۰ و ۴٫۵، ارائه می دهد. اثر افزودن آلکوفین دوز بر روی تجزیه و تحلیل گرماسنجی انجام گردیده که در آن نه تنها از دست دادن و تجزیه محصولات هیدراتاسیون از بتن با توجه به درجه حرارت یافت شد، بلکه همچنین آب متصل (٪) و هیدروکسید کلسیم (٪) موجود در بتن تعیین گردید. از پراش اشعه ایکس مشاهده شد که آلکوفین تشکیل هیدرات کلسیم را در بتن و تجزیه و تحلیل میکروسکوپ الکترونی اسکن ( SEM) شان داد که تشکیل هیدراتهای اترینگیت و هیدرات سیلیکات کلسیم در حفره میکی که بتن را تسریع می کنند. بنابراین، از این مطالعه نتیجه گرفته شده است که آلکوفین را می توان به عنوان یک جایگزین مناسب برای سیمان در بتن معمولی با توجه به اثرات مثبت آن در بهبود ارتقاء سیمان و یک محصول سازگار با محیط زیست مورد استفاده قرار می گیرد.

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